

ISLAMIC UNIVERSITY OF SCIENCE & TECHNOLOGY
DEPARTMENT OF CHEMISTRY



M Sc. (3rd Semester) (Mid term)
Course Title: Inorganic spectroscopy (PCH-DCE-306)
Rollno:

Max. Marks: 30
Time: 40min

- Q1. The low temperature ^{19}F NMR spectrum of IF_5 molecule in solution should exhibit which of the following patterns (ignore any magnetic coupling effects to the Iodine nucleus)
- One singlet
 - One singlet and one Quintet
 - One doublet and one Quintet
 - One quartet and one triplet
- Q2. The ^{31}P NMR spectrum of $\text{PF}_4\text{N}(\text{CH}_3)_2$ at room temperature and low temperature (173K) respectively shows (assume that N & H do not couple)
- Triplet and Quintet
 - Quintet and Triplet of Triplet
 - Quintet and Triplet
 - Triplet and Triplet of Triplet
- Q3. In NMR spectrum, nuclei that are located near to electronegative atom tend to be relative to nuclei that are not near to electronegative atoms.
- Split
 - Have a more intense signal
 - Shielded
 - Deshielded
- Q4. The NMR experiment consists of measuring the energy associated with what physical phenomenon?
- Excitation of a core electron from an atom
 - Flipping the magnetic spin vector of a nucleus in a strong magnetic field
 - Activation of molecular vibration
 - Promoting of an electron from the HOMO to the LUMO
- Q5. Which of the following factor does not affect the chemical shifts
- Inductive effect
 - Anisotropic effect
 - Concentration
 - Hydrogen bonding
- Q6. The ^{19}F NMR spectrum of ClF_3 shows
- Doublet and triplet for a T-shaped structure
 - Singlet for a trigonal planar structure
 - Singlet for a trigonal pyramidal structure
 - Doublet and singlet for a T-shaped structure
- Q7. The ^{31}P NMR spectrum of P_4S_3 consists of:
- A doublet and a quartet
 - Two doublet
 - A doublet and a triplet
 - A singlet
- Q8. A borane (X) is reacted with NH_4 to give a solution of borohydride (Y). The ^{11}B NMR spectrum of Y consists of a triplet and a quintet. The borane X is:
- B_2H_6
 - B_3H_9
 - B_4H_8
 - B_5H_9
- Q9. In the ^{19}F NMR spectrum of PF_5 , the number of signals and multiplicity at room temperature are:
- One, doublet
 - One, singlet
 - Two, singlet
 - Two, doublet
- Q10. The signal of paramagnetic substances.....
- Broad and have wide chemical shift range
 - Sharp and have wide chemical shift range
 - Broad and have small chemical shift range
 - Sharp and have small chemical shift range

Signature of Student

Signature of Invigilator

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- Q11. Main role of MRI contrast agent
- Increases the T1 relaxation time of proton inside the tissue
 - Increases the T2 relaxation time of proton inside the tissue
 - Decrease the T1 relaxation time of proton inside the tissue
 - Decrease the T2 relaxation time of proton inside the tissue
- Q12. The multiplicity of the signal in ³¹P NMR spectrum in PD₃ is
- Singlet
 - Septet with unequal intensities
 - Quartet with equal intensities
 - Septet with equal intensities
- Q13. [B₃H₈]⁻ is fluxional on the NMR spectroscopic timescale. In the ¹¹B and ¹¹B{¹H} NMR spectra respectively, you would observe:
- A singlet and an octet
 - An octet and a singlet
 - A singlet and a nonet
 - A nonet and a singlet
- Q14. [(η³-C₃H₅)Mn(CO)₄] shows fluxional behaviour. The ¹H NMR spectrum of this compound when it is in the non fluxional state shows:
- One signal
 - Two signals in the intensity ratio of 4:1
 - Three signals in the intensity ratio of 2:2:1
 - Five signals of equal intensity.
- Q15. The ligand that are fluxional in [(η⁵-C₅H₅)(η¹-C₅H₅)Fe(CO)₂] in the temperature range 221-298K, are
- η⁵-C₅H₅
 - η¹-C₅H₅
 - η⁵-C₅H₅ and CO
 - η¹-C₅H₅ and CO
- Q16. A major advantage of MRI is:
- The ease with which equipment is updated or replaced
 - Its relatively low cost, compare to CT scans
 - Does not require specialized room
 - The ability to reposition the cross section through the body without repositioning the patient.
- Q17. What is the relative area of each peak in a quartet spin spin splitting pattern?
- 1:2:2:1
 - 1:2:1
 - 1:4:4:1
 - 1:3:3:1
- Q18. The reference material used in BNMR spectroscopy is:
- BF₃.OEt₂
 - BF₃.OBu₂
 - BF₃.OEt₃
 - None of these
- Q19. The multinuclear NMR spectroscopy depends on
- Sensitivity
 - Isotopic abundance
 - Both (a) and (b)
 - None of these
- Q20. Paramagnetic shifts observed in
- Organic radicals
 - Complexes of transition metals
 - Complexes of lanthanides and actinides
 - All of the above.

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